

P/047/61/012/001/001/002  
D221/D306

24.69<sup>00</sup>  
AUTHORS:

TITLE:

Czyżewski, Oleg, and Hołyński, Roman  
Multiple generation of particles in nucleon-nucleon  
and  $\pi$ -meson-nucleon collisions, within certain accele-  
ration energy ranges

PERIODICAL: Postępy fizyki, v. 12, no. 1, 1961, 71 - 87

TEXT: Work done on inelastic n-n and  $\pi$ -n collision in Western and Soviet-bloc establishments is reviewed, and a theoretical explanation is sought for the results obtained, such as multiplicity of mesons generated, their energetic and angular distribution and coefficients of inelasticity of collisions. It is shown that Fermi's model is insufficient, while an isobaric model agrees with experimental results only for low multiplicity of produced mesons. For proton-nucleon collisions the following works are discussed: N.P. Bogachev, I.M. Gramenitskiy, V.B. Lubimov, Y.P. Merekov, M.Y. Podgoretskiy, V.N. Sidorov, and D. Tuvdendorzh (Ref. 1: Zh. Exper.

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Multiple generation of particles ...

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Teor. Fiz., 37, 1225, 1959), (Ref. 2: Wan-Szu-Fen, T. Wiszki, I.M. Gramenitskiy, V.G. Grishin, N. Dolkhashav, R.M. Lebedev, A.A. Nomofilov, M.Y. Podgoretskiy, V.N. Streltsov, Preprint - Dubna, 1960), (Ref. 3: R.R. Daniel, N. Kameswara Rao, P.K. Malhotra, Y. Tsusuki, Preprint, Bombay, 1959), (Ref. 4: R. Kalbach, J. Lord, C. Tsao, Phys. Rev., 113, 325, 1959). In the first of the above works emulsion technique was used and energy of proton beam was 9 BeV. The second was a continuation of the first. The latter two both deal with the same problem also using emulsion technique with the proton beam of energy of 6.2 BeV. For pion-nucleon collision at the energy of the order of 1 BeV, the isobaric model gives a very good agreement with experimental results but it fails at energies of several BeV. Results of W.D. Walker (Ref. 10: Phys. Rev., 108, 872, 1957) with  $\pi$ -mesons of 4.5 BeV and of G. Maenchen, W.B. Fowler, W.M. Powell, R.W. Wright (Ref. 11: Phys. Rev., 108, 850, 1957) with  $\pi$ -mesons of about 5 BeV lead to the assumption of  $\pi$ - $\pi$  collision, where  $\pi$ -meson collides with another  $\pi$ -meson in the meson cloud of the nucleon.

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Work by V.A. Byelakov, Wan-Szu-Feh, W.W. Glagolev, N. Dolkhashav, P.M. Lebedev, N.N. Melnikova, V.A. Nikitin, V. Petrzhelka, V.A. Svipidov, M. Suk, K.D. Tolstov (Ref. 14: Preprint, Dubna 1960) with  $\pi$ -mesons of the energy of 6.8 BeV and J. Bartke (Ref. 17: Komunikat prywaty, 1960) who used  $\pi$ -mesons of 16 BeV/C also both lead to the assumption that at least a part of mesons is produced by  $\pi$ - $\pi$  interaction. Finally a pion-nucleus interaction is investigated by H.H. Aly, J.G.M. Duthie, C.M. Fisher (Ref. 18: Phil. Mag., 4, 993, 1959) where a stack of nuclear emulsion was irradiated with  $\pi^-$  of 4.5 BeV. Collisions with heavy nuclei produced angular distribution different to those obtained from collision with lighter nuclei, protons and by grazing collisions. The authors suggest that in the first case a greater number of nucleons is involved in the collision with  $\pi$ -meson, but it is pointed out that a cascade-type mechanism would lead to the same results. In view of the inability of statistical theory to explain the results cited above, two alternatives are mentioned. The first is the isobaric model which may be still further improved, according to I. Tamm (Ref. 19: Materialy

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IX Konferentsii Fiziki Vysokikh Energii, Kiyev, 1959) and the se-  
cond is the introduction of inner structure of the nucleon which  
would certainly take part in multiple meson generation and  $\pi$ -heavy  
nucleus collision. There are 31 graphs, 3 tables, 2 photographs and  
19 references, 7 Soviet-bloc and 12 non-Soviet-bloc. The references  
to the four most recent English-language publications read as fol-  
lows: R.R. Daniel, N. Kameswara Rao, P.K. Malhotra, Y. Tsusuki,  
Preprint, Bombay, 1959; R. Kalbach, J. Lord, C. Tsao, Phys. Rev.,  
113, 325, 1959; J.G.M. Duthie, H.H. Aly, C.M. Fisher, Phil. Mag.,  
4, 993, 1959; R.B. Sternheimer, S.J. Lindenbaum, Phys. Rev., 109,  
1723, 1958.

D221/D506

ASSOCIATION: Zakład VI instytutu badań jądrowych, Kraków (VIth In-  
stitute of Nuclear Research, Cracow)

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P/045/61/020/004/004/004  
B133/B215

## AUTHORS:

Bartke, J., Czachowska, Z., Holynski, R., Rybicki, K.

## TITLE:

Some examples of interaction of protons of very high energy  
with heavy nuclei of photographic emulsions

## PERIODICAL:

Acta Physica Polonica, v. 20, no. 4, 1961, 331-339

TEXT: Three stars produced in collisions with nuclei of a photographic emulsion are described. Although they were probably produced in collisions of nucleons with heavy nuclei, they show double maximum angular distributions in contradiction with the hydrodynamic model. Star I: 26 + 47p; star II: 18 + 41p; and star III: 15 + 78p. Stars I and II have been found in a stack of Ilford G5 emulsions irradiated in the  $\mu$  valley in 1957, and star III was found in an NIKFI-R emulsion stack irradiated near Moscow in 1958. Target diagrams at distances of 600, 1000, and 1400 from the primary interaction enabled the authors to distinguish between the tracks from the secondary interaction and those from the primary event. The angles between the tracks of all primary particles and the star axis were measured. From these angles, the Lorentz factor of the system can

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Some examples of interaction of ...

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be determined as  $\gamma_c = \log \cot \theta_i$ . The primary energy  $E_p = 2\gamma_c^2 Mc^2$  is obtained for nucleon - nucleon collisions in the laboratory system. For nucleon - nucleus collisions  $E_p = 2n\gamma_c^2 Mc^2$ , is obtained by using the tunnel model.  $n$  is the number of particles in the tunnel. A measure for the anisotropy in the angular distribution is the dispersion

$$\sigma = \sqrt{\frac{\sum (\log \tan \theta_i - \bar{\log} \tan \theta)^2}{n-1}} \quad (A).$$

The values of  $E_p$ ,  $\gamma_c$ , and  $\sigma$  for the events described are presented in Table II

jet	type	$\gamma_c$	$E_p = 2\gamma_c^2 Mc^2$	$E_p = 2n\gamma_c^2 Mc^2$	$\sigma$
I	26+47p	83.1	$1.3 \times 10^{13}$ eV	$5.2 \times 10^{13}$ eV	1.25
II	18+41p	58.1	$6.4 \times 10^{12}$ eV	$2.5 \times 10^{13}$ eV	1.10
III	15+78p	15.3	$4.4 \times 10^{11}$ eV	$1.8 \times 10^{12}$ eV	0.71

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All differential angular distributions show strong anisotropy and two maxima which corresponds to a plateau in the integral distribution, as may be seen from Fig. 2a. The angular distributions of gray and black tracks do not deviate significantly from the isotropic distribution. Feinberg (Feinberg, E. L., Uspekhi fiz. Nauk, 70, 333, (1960). (Presented also at the Moscow and Kiev conferences)) has expressed the opinion that there are two types of nucleon-nucleon collisions, namely, head-on collisions and peripheral collisions. The hydrodynamical model can be applied only to the first type which is obviously present (Milekhin, G. A., Zh. eksper. teor. Fiz., 35, 1185, (1958)). According to this theory, the differential angular distribution can be well described by a Gaussian curve which is compared in Fig. 4 with the values obtained. In a paper by Gierula et al. (Gierula, J., Miesowicz, M., Zielinski, P., Acta phys. Polon., 19, 119 (1960)) where the three stars under consideration have been referred to as 171K, 168K, and 200K, respectively, a measure has been defined for the deviation predicted by the two-center model. The deviation is calculated in these units and according to the Kolmogorov-Smirnov test which, in the authors' view, cannot be applied here (Smirnow, H., Recueil Mathematique N.S.6, 3 (1959)). It is concluded that the experimental facts do not con-

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firm the hydrodynamical model, whereas a two-center model describes the phenomena very well if the "central" collisions take place in a very small "core". The authors thank Professors M. Miesowicz and J. Gierula. There are 4 figures, 3 tables, and 11 references: 4 Soviet-bloc and 7 non-Soviet-bloc.

ASSOCIATION: Cosmic Ray Department of the Institute of Nuclear Research  
Cracow - Poland

SUBMITTED: October 24, 1960

Card 4/6

GIERUIA, J.; HOLYNSKI, R.; MIESOWICZ, M.

Interactions of nucleons with heavy nuclei of photographic emulsions at energies higher than  $10^{12}$  eV. Acta physica Pol 22 no.4:329-334 0 '62.

1. Institute of Nuclear Research, Laboratory of High Energy Physics, Krakow Department, Krakow, and 2d Department of Physics, Academy of Mining and Metallurgy, Krakow.

HOLYST, Brunon

Estimation of the body height from the length of the foot. Arch.med.  
sad., Warszawa 6:158-173 1955.

1. Z Zakladu Kryminologii Uniwersytetu Lodzkiego. Kierownik: prof.  
dr P. Horoszowski.

(BODY HEIGHT

(FOOT  
determ. from length of foot prints in forensic med.)  
foot print length as basis for determ. of body height  
in forensic med.)

"APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0

HOLYST, Jerzy

Echoencephalography. Neurol., neurochir., psychiat. Pol. 14  
no.3:447-453 My-Je '64

1. Z Kliniki Neurologicznej Akademii Medycznej we Wrocławiu  
(Kierownik: prof. dr. R. Arend).

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

HOLYST, Jerzy

Intracranial hemorrhage in hemophilia. Neurol. neurochir.  
psychiat. pol. 13 no. 5:601-606 '63.

l. Z Kliniki Neurologicznej AM we Wrocławiu. Kierownik: prof.  
dr. R.Arend.

HOLYST, Jerzy

Echo-encephalography (ultrasonic-encephalography). Pol. tyg.lek,  
18 no. 47:1749-1752 '18.N°63

1. Z Kliniki Neurologicznej AM we Wrocławiu; kierownik: prof.  
dr. Rudolf Arend.

HOLIST, Jerzy  
SURNAME (in caps); Given Name

Country: Poland

Academic Degrees:

Affiliation: Neurological Clinic, School of Medicine (Akademia Medyczna)  
Wroclaw; Director: R. AREND, Prof dr med

Source: Warsaw, Przeglad Lekarski, No 4, 1961, pp 187-189

Data: "A Dissecting Aneurysm on the Main Abdominal Artery with  
Contour Calcification Ascertained Intravitally."

HOLYST, Jerzy

Unusual form of phacomatosis. Neurologia etc. polska 11 no. 6: 343-  
846 '61.

1. Z Kliniki Neurologicznej AM we Wrocławiu Kierownik: prof. dr  
R.Arend.  
(ABNORMALITIES)

RUDKOWSKA, Anna; KRAUSE, Krystyna; HOLYST, Jarey

Electroencephalographic changes during the course of tofranil therapy of depressive states. Neurologia etc. polska 11 no.2: 241-250 Mr-Ap '61.

1. Z Kliniki Neurologicznej AM we Wrocławiu Kierownik: prof. dr R. Arend i z Kliniki Psychiatrycznej AM we Wrocławiu Kierownik: doc. dr M. Demianowska.

(DEPRESSION ther) (PSYCHOPHARMACOLOGY)  
(ELECTROENCEPHALOGRAPHY)

HOLYST, Jerzy; KRAUSE, Krystyna

Neurological and psychiatric syndromes in thallium poisoning.  
Polski tygod. lek. 16 no.9:337-340 27 F '61.

1. Z Kliniki Neurologicznej A.M. we Wrocławiu; kierownik: prof.  
dr Rudolf Arend i z Kliniki Psychiatrycznej A.M. we Wrocławiu;  
kierownik: doc. dr Maria Demianowska.

(THALLIUM toxicol) (NEUROLOGICAL MANIFESTATIONS)

TOKARZ, Feliks; HOLYST, Jerzy

Tumors in the area of the foramen magnum. Neurol. neurochir.  
psychiat. Pol. 14 no.1:55-62 Ja-F '64.

1. Z Kliniki Neurochirurgii Akademii Medycznej w Poznaniu  
(Kierownik: doc. dr. med. H. Powiertowski).

HOLYST, Jerzy; KOTICKI, Andrzej; KRAUSE, Krystyna

Foreign bodies in the brain as a result of self-mutilation.  
Neurochir., Psychiat. Vol. 14 no.4:581-588 Jl.-Ag 64

1. Z Kliniki Neurochirurgii Akademii Medycznej w Poznaniu  
(Kierownik: doc. dr. H. Powiertowski) i z Klinik Psychiatrycznej  
Akademii Medycznej we Wrocławiu (Kierownik: doc. dr. M. Lemiecowska).

"APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0

TOKARZ, Feliks; HGLYŚT, Jerzy; GRADZKI, Janusz

Anomaly of Galen's vein. Neurol., neurochir., psychiat. Pol.  
14 no.3:541-543 My-Je '64

1. Z Kliniki Neurochirurgii Akademii Medycznej w Poznaniu  
(Kierownik: doc. dr. H. Powiertowski).

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

KRAUSE, Krystyna; HOLYST, Jerzy

Diagnostic difficulties in a case of subdural hematoma. *Neuroloj., neurochir. Psychiat. Pol.* 14 no.3 549-552 Mar-Ju '64

1. Z Kliniki Psychiatrycznej Akademii Medycznej we Wrocławiu (Kierownik: doc. dr. M. Demianowska) i z Kliniki Neurochirurgii Akademii Medycznej w Poznaniu (Kierownik: doc. dr. H. Powiertowski).

RUDNICKI, Stanislaw; KRAUSE, Krystyna; HOLYST, Jersy

Polysymptomatic neuro-psychiatric syndrome as a sequel of an anomaly  
of the anterior part of the circle of Willis. Neurologia etc., polska  
12 no.2:265-273 '62.

1. z Kliniki Neurochirurgii AM w Warszawie Kierownik: prof. dr J. Chorobski  
z Kliniki Psychiatrycznej we Wrocławiu Kierownik: doc. dr M. Demianowska  
i z Kliniki Neurologicznej we Wrocławiu Kierownik: prof. dr R. Arend.  
(CEREBRAL ARTERIES abnorm) (NEUROLOGICAL MANIFESTATIONS)

HOLYST, J.

"Etiopathogenic studies on 323 brain strokes" by L. Iwanowski. Reviewed by J. Holyst. Neurol neurochir psych 12 no.2:310 Mr-Ap '62.

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"APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0

HOLYST, J.

"History and development of neurology in Poland." Neurol neurochir psych 12 no.2:310 Mr-Apr '62.

\*

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

HOLYST, J.

"Myelopathy resulting from tritolyphosphorane poisoning" by H. Geofroy and others. Reviewed by J. Holyst. Neurolog neurochir psych 12 no.2:312 Mr-Ap '62.

\*

HOLYST, J.

"The possible role of vaccines and sera in the pathogenesis of multiple sclerosis" by G. Palffy, F.T. Merei. Reviewed by J. Holyst. Neurol neurochir psych 12 no.2:312-313 Mr-4p '62.

KRAUSE, Krystyna; HOLYST, Jerzy

Psychic disturbances after the reactivation of the heart. Neurol  
neurochir psych 12 no.3:401-408 My-Je '62.

1. Klinika Psychiatryczna, Akademia Medyczna, Wrocław (Kierownik:  
doc. dr M. Demianowska) i Klinika Neurologiczna, Akademia Medyczna,  
Wrocław, Kraszewskiego 25. (Kierownik: prof. dr R. Arend).

HOLYST, J.

"The spinal cord. Basic aspects and surgical considerations"  
by G.Austin. Reviewed by J. Holyst. Neurol neurochir psych  
12 no.4:631-632 Jl-Ag '62.

HOLYST, J.

"A functional approach to neuroanatomy" by E.L.House, B.Pansky.  
Reviewed by J.Holyst. Neurol neurochir psych 12 no.4:632-633  
Jl-Ag '62.

4

HOLYST, J.

"The effect of advancing age upon the human spinal cord" by L.R.  
Morrison, S.Cobb, W.Bauer. Reviewed by J.Holyst. Neurol  
neurochir psych 12 no.6:933-934 N-D '62.

"APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0

HOLYST, J.

"Intraspinal tumors of childhood" by R.W.Rand, C.W.Rand. Reviewed  
by J.Holyst. Neurol neurochir psych 12 no.6:934 N.D '62.

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

HOLYST, Jerzy; KRZYSZTON, Zofia

Dermatomyositis simulating bulbar syndrome. Pol. tyg. lek. 17 no.13:  
484-486 26 Mr '62.

l. Z Kliniki Neurologicznej AM we Wrocławiu; Kierownik prof. dr  
Rudolf Arend.

(DERMATOMYOSITIS diag)

POLAND

Jerzy HOLYST, Witold DOLATA and Tadeusz ORLOWSKI, Neurology Clinic of College of Medicine (Klinika Neurologiczna Akademii Medycznej), Head (kierownik) Prof Dr k. AREND; and Department of General Surgery, Regional Army Hospital (Oddzial Chirurgii Ogolnej Wojskowego Szpitala Okregowego), Head Physician (Ordynator) physician (lekarz) T. ORLOWSKI, Wrocław.

"Cerebral Complications and Changes after Cardiac Arrest."

Krakow, Przeglad Lekarski, Vol 18/Ser 2, No 11, 1962; pp 428-430.

Abstract [English summary modified]: Description of four cases in which circulation was arrested for 1, 4, 6 and 10 minutes respectively; it was then restored in all but all four eventually died ~36, 29, 17 and 5 hours later with pyrexia, pulmonary edema, respiratory center failure. Direct cardiac massage, hibernation, injection of oxygenated blood under pressure directly into carotid arteries are advocated as probably the most promising therapeutic method in such cases. Three Polish and 14 Western references.

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HRYNKIEWICZ, Leon and HOLYST, Jerzy, Psychiatric Clinic (Klinika Psychiatryczna, (Director: Docent, Dr. M. DENIA-NOWSKA) and the Neurological Clinic (Klinika Neurologiczna) (Director: Prof. Dr. R. AREND), both of the AW [Akademia Medyczna, Medical Academy] in Wrocław

"Unusual Complication of the Anticoal-Alcohol Reaction.  
Report of Two Cases."

Warsaw, Polski Tygodnik Lekarski, Vol 18, No 3, 14 Jan 53,  
pp 93-95.

Abstract: [Authors' English summary modified] Two unusual cases are described. In one, with a typical onset, a syndrome appeared like in cerebral stroke with hemiparesis, symptoms of atropine poisoning and mental disturbances. All signs disappeared after two days. In the second case, of alcohol poisoning, severe consciousness disturbances and fatal circulatory and respiratory troubles developed. Of the 16 references, five are Western, and 11 Eastern.

1/1

POLAND

HOLYST, Jerzy, Neurological Clinic (Klinika Neurologiczna),  
AM [Akademia Medyczna, Medical Academy] in Wroclaw (Director:  
Prof. Dr. Rudolf AREND)

"Echo-Encephalography in Study of Organic Intracranial Changes."  
Warsaw, Polski Tygodnik Lekarski, Vol 18, No 30, 22 Jul 63,  
pp 1109-1111

Abstract: Review article discussing the principle and apparatus used in, and the application of supersound in medicine for both diagnostic and therapeutic purposes, the effect of supersonic waves on brain tissue, and in greater length the use of ultrasonic techniques and echo-encephalography for the study of the structure of the brain. There are two illustrations of an echo encephalogram of normal brain and explanation of the I, M, and S areas. There are 30 references, of which two (2) are in Polish, one (1) in French, and the others in English.

1/1

HOLYST, Jerzy

Echo-encephalography in the diagnosis of intracranial hematomas.  
Pol. tyg. lek. 19 no.48:1835-1837 30 N'64.

1. Z Kliniki Neurochirurgicznej Akademii Medycznej w Poznaniu  
(kierownik: doc. dr. H. Powiertowski).

HOLYST, Jerzy; KRAUSE, Krystyna

Clinico-statistical evaluation of multiple sclerosis in Lower  
Silesia. Pol. tyg. lek. 20 no.10:337-340 8 Mr '65

l. Z Kliniki Neurologicznej Akademii Medycznej we Wrocławiu  
(Kierownik: prof. dr. Rudolf Arend).

WOLNST. Jerzy; GRADZKI, Janusz

Use of subtraction in neuroradiology. Pol. tyg. lek. 19 no.7;  
252-255 10 F '64.

1. Z Kliniki Neurochirurgii Akademii Medycznej w Poznaniu  
(kierownik: doc. dr Hieronim Powiertowski).

TOKARZ, Feliks; HOLYST, Jerzy

Surgical therapy of aneurysms of communicating arteries. Neurol.,  
neurochir., psychiat. Pol. 15 no.1:135-143 Ja-F'65.

l. Z Kliniki Neurochirurgii Akademii Medycznej w Poznaniu  
(Kierownik: doc. dr. med. H. Powiertowski).

TOKARZ, Feliks ; HOLYST, Jerzy; STRZYZEWSKI, Włodzimierz

Intracranial management of an arteriovenous angiomyoma of the central cerebral region. Neurol., neurochir., psychiat. Pol. 15 no.1:191-193 Ja-F'65.

1. Z Kliniki Neurochirurgii Akademii Medycznej w Poznaniu (Kierownik: doc. dr. H. Powiertowski) i z Oddziału Neurologicznego Szpitala im. Strusia w Poznaniu (Kierownik: dr. T. Frackowiak).

HOLYST, Jerzy

Echoencephalography. Postepy hig. med. dosw. 19 no.2:273-302  
Mr-Ap '65.

1. Z Kliniki Neurologicznej AM we Wrocławiu (Kierownik: prof.  
dr. R. Arend).

HOLZBECHER, K., inz.

Activities of the Third Subcommission of the International Gas  
Union in Czechoslovakia. Paliva 44 no.8:258-259 Ag '64.

"APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0

HOLZBECHER, K.; MUSIL, J.

Radiant burners. Prace Ust paliv no. 5:34-93 '62.

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

HOLZBECHER, K.

F

3847. USE OF PROPANE OR BUTANE FOR PEAK LOAD PERIODS IN GAS INDUSTRY.  
Slivo, V. and Holzbecher, K. (Paliva (Fuel), Feb. 1951, vol. 31, 23-33).  
A description is given of a propane-air cracking gas producer. This  
producer can also be used, with slight modifications, for butane-air or  
propane-oxygen (eventually butane-oxygen) cracking. The authors attribute  
great importance to the cracking of hydrocarbon-oxygen mixtures in order to  
achieve a high production. (L).

HOLZBECHER, K.

"Utilisation of Infrared Gas Burners in Industry and Agriculture," p. 86.  
(Paliva, Vol. 33, No. 4, Apr. 1953, Praha.)

SO: Monthly List of East European Accessions, Vol. 2, No. 9, Library of Congress, September  
1953, Unclassified.

"APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0

HOLZRECHER, K.

"Survey of Contemporary Development in the Production of High-Capacity Gas Burners." p. 26,  
Praha, Vol. 34, no. 2, Feb. 1954.

SO: East European Accessions List, Vol. 3, No. 9, September 1954, Lib. of Congress

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

HOLZBECHER, K.

"Heating industrial enterprises with infrared gas burners." p. 705.

STROJIRENSTVI. (MINISTERSTVO TEZKEHO STROJIRENSTVI, MINISTERSTVO PRESNEHO STROJIRENSTVI A MINISTERSTVO AUTOMOBILOVEMU PRUMYSLU A ZEMEDELSKYCH STROJU.)  
Praha, Czechoslovakia, Vol. 5, no. 9, Sept. 1955.

The advantages are outlined and an illustrated description is given of flame-less gas burners, consisting mainly of porous plates, which have been designed by the Research Institute of the Gas Industry and used in their hall. (L)/

Monthly List of East European Accessions (EEAI), LC, Vol. 8, No. 9, September 1959.  
Uncl.

RECHER  
HOLZBERGER, K.

Industrial gas appliances. p. 236

PALIVA.(Ministerstvo paliv a Ceskoslovenska vedecka technicka spolecnost pro  
vyuziti paliv pri Ceskoslovenske akademii ved) Praha, Czechoslovakia, Vol. 39,  
no. 7, July 1959.

Monthly list of East European Accessions (EEAI) LC, Vol. 8, No. 11,  
November 1959.

uncl.

"APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0

HOLZBECHER, K., ins.

Meeting of the Subcommittee on Use of Gas in Industries and Municipal Enterprises of the International Gas Union. Paliva 42 no.12:374 D '62.

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

HOLZBECHER, Kristian, inz.

Possibility of economical consumption of gas in industry.  
Energetika Cz 13 no.1:22-23 Ja '63.

1. Ustav pro vyzkum paliv, Bechovice.

MINKO, V., inz.; HOLZBECHER, K., inz.

Radiation burners for technological processes with temperatures  
up to 1400° C. Paliva 44 no.2:41-45 F'64.

1. Ustav pro vyzkum paliv, Bechovice.

"APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0

HOLZBECHER, K.

Use of gas for radiant heating. Paliva 44 no.5/6:194-198  
My-Je '64.

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

Determination of uranium in the presence of cerium (III), thallium (I), silver, lead, copper, mercury (II), and potassium ferricyanide by means of bathophenanthroline. V. Novotna and Z. Holoubek. Collection Czechoslov. Chem. Commun., 14, 40-58 (1949) (in French); cf. C.A. 32, 57232; 33, 57097; 34, 57811. —The usual method of detg. U with bathophenanthroline is applicable in the presence of Ce (III) or Tl (I). Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub> is used to prevent pptn. of Ag, Pb, or Cu if present. To prevent pptn. of Hg (II) if present excess Cl<sup>-</sup> is used to form HgCl<sub>4</sub><sup>2-</sup>. The sepn. of Fe (II) from U cannot be carried out by complexing the Fe as cyanide and oxidizing to Fe(CN)<sub>6</sub><sup>4-</sup> since a blue compd. is absorbed by the ppt. of uranyl bisoxime.

Peter M. Bernays

Metallic salts of the  $\beta$ -semicarbazone of hattin and its derivatives. V. Horváth and Z. Höglinger. *Collection Czechoslov. Chem. Commun.*, 16, 116-130 (1949) (in French); cf. C. A. 42, 15236.—A 0.5% soln. of the  $\beta$ -semicarbazone of hattin (I) (see Marchlewski, *Ber.*, 29, 1012 (1896)) in 10% alc. gives an orange or yellow ppt. with  $\text{Ag}^+$ ,  $\text{Hg}^{2+}$ ,  $\text{Hg}^{2+}$ , and  $\text{Bi}^{3+}$ ; brown, but no ppt. with 15 other cations tested;  $\text{Fe}^{2+}$  and  $\text{UO}_2^{2+}$  color the alc. soln. orange-brown. A soln. of 10 g. I and 2 g. NaOH in 1 l. 35% ROH gives an orange-yellow ppt. with  $\text{Ag}^+$ ,  $\text{Pb}^2+$ ,  $\text{Ti}^{2+}$ ,  $\text{Hg}^{2+}$ , and  $\text{Bi}^{3+}$ , brown-green with  $\text{Hg}^{2+}$ , yellow-brown with  $\text{Cu}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Mn}^{2+}$ , and  $\text{Zn}^{2+}$ , brown with  $\text{Fe}^{2+}$ , yellow with  $\text{Ca}$  after 1 hr., and yellow with  $\text{Ba}^{2+}$  and  $\text{Sr}^{2+}$ , while  $\text{Zr}^{4+}$  and  $\text{Tb}^{3+}$  faintly color the soln. orange and brown, resp.; the following salts were analyzed:  $\text{Hg}(\text{C}_6\text{H}_5\text{N}_3\text{O}_2)_2$ ,  $\text{TiCl}_3\text{H}_2\text{N}_3\text{O}_2$ ,  $\text{Sr}(\text{C}_6\text{H}_5\text{N}_3\text{O}_2)_2$ ,  $\text{Ba}(\text{C}_6\text{H}_5\text{N}_3\text{O}_2)_2$ , and  $(\text{PbCl}_2\text{H}_2\text{N}_3\text{O}_2)_2$ . A soln. of 10 g. of the  $\beta$ -semicarbazone of *N*-methylhattin and 2.25 g. NaOH in 1 l. 35% alc. gives an orange or yellow ppt. with  $\text{Ag}^+$ ,  $\text{Hg}^{2+}$ , and  $\text{Pb}^{2+}$ , green with  $\text{Hg}^{2+}$ , brown-green with  $\text{Cu}^{2+}$ ,  $\text{Cd}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Mn}^{2+}$ ,  $\text{Zn}^{2+}$ , and  $\text{Fe}^{2+}$ , yellow with  $\text{Sr}^{2+}$  and  $\text{Ba}^{2+}$ , and yellow with  $\text{Ca}$  after several hrs.;  $\text{Pb}(\text{C}_6\text{H}_5\text{N}_3\text{O}_2)_2$  was analyzed. A soln. of 10 g. of the  $\beta$ -semicarbazone of *N*-benzylhattin and 2 g. NaOH in 1 l. 60% ROH gives a

brown or yellow-orange (ppt. with  $\text{Ag}^+$ ,  $\text{Hg}^{2+}$ ,  $\text{Hg}^{2+}$ ,  $\text{Pb}^{2+}$ , and  $\text{Tl}^{+}$ ), yellow, brown, or brown-green with  $\text{Cu}$ ,  $\text{Cd}$ ,  $\text{Ni}$ ,  $\text{Co}$ ,  $\text{Zn}$ ,  $\text{Mn}$ , and  $\text{Fe}^{2+}$ , and yellow with  $\text{Cr}$ ,  $\text{Se}$ , and  $\text{Ba}$ .  $\text{Pb}(\text{C}_{12}\text{H}_{22}\text{NO}_3)_2$  and  $\text{Mn}(\text{C}_{12}\text{H}_{22}\text{NO}_3)_2$  were also analyzed. The action of dil. acids,  $\text{NaOH}$ ,  $\text{KOH}$ , and excess reagent on the various salts is noted. P. M. Dugney

ASA-31A METALLURGICAL LITERATURE CLASSIFICATION

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CIA-RDP86-00513R000618120003-0"

(A)

Metallic salts of the 3-thiosemicarbazones of isatin and its derivatives. V. Hovorka and Z. Holzbecher. *Colloque sur les Cétones Thiosemicarbazones*. Chem. Commun., 14, 248-252 (1979) (in French); cf. C.A. 83, 8079x. A 0.5% soln. of the 3-thiosemicarbazone of isatin (I) in 90% alc. gives a yellow to orange ppt. with Ag, Hg<sup>++</sup>, and Bi, brown-green with Hg<sup>+</sup>, but no ppt. with Cd, Mn, Zn, and Fe<sup>++</sup>, while Cu<sup>++</sup>, Ni, and Co color the soln. brown, and Fe<sup>+++</sup>, Sb<sup>+++</sup>, and UO<sub>4</sub><sup>++</sup> color it orange; the addn. of AcONa gives a yellow to brown ppt. with Cd, Ni, Cu, Mn, Cu<sup>++</sup>, and Zn, and dark green with Fe<sup>++</sup>. A soln. of 10 g. I and 2.1 g. NaOH in 11.50% EtOH gives a yellow to orange ppt. with Ag, Hg<sup>+</sup>, Hg<sup>++</sup>, and Bi, and orange with Pb and K; the following salts were analyzed: Hg(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Ph(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Ti(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Ni(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Cu(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Zn(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>. A 0.5% soln. of the 3-thiosemicarbazone of 1-methylisatin (II) in Me<sub>2</sub>CO gives an orange ppt. with Ag and Hg<sup>++</sup>, and a brown-green ppt. with Hg<sup>+</sup>, while Fe<sup>++</sup> colors the soln. dark brown, and UO<sub>4</sub><sup>++</sup> orange; ppts. are obtained with Cu, Cd, Ni, Co, Zn, and Mn only after the addn. of AcONa. A soln. of 10 g. II and 3 g. NaOH in 11.70% EtOH gives a brown-green ppt. with Hg<sup>+</sup> and Ag, orange with Hg<sup>++</sup>, Pb, Bi, and Bi, brown, orange, or yellow with Cu, Cd, Ni, Co, Zn, Mn, and dark green with Fe<sup>++</sup>; the following salts were analyzed: Hg(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Ph(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Ti(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Ni(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, Co(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>, and Zn(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub>. A soln. of 10 g. of the 3-thiosemicarbazone of 1-benzylisatin and 2 g. NaOH in 1 l. of 73% Me<sub>2</sub>CO gives yellow and brown to red-brown ppts. with Ag, Pb, Hg<sup>++</sup>, Ti, Bi, Cu, Cd, Ni, Co, Mn, and Zn, and a dark green ppt. with Hg<sup>+</sup> and Fe; Pb(C<sub>11</sub>H<sub>10</sub>N<sub>2</sub>OS)<sub>2</sub> was analyzed.

The action of dil. acids, NaOH, NH<sub>4</sub>OH, and excess reagents on the various salts is noted. P. M. Downey

*CA*

Microchemical tests for sulfate, thiosulfate, sulfide, hydrosulfide, and formaldehyde. V. Hovorka and G. Holzbecher (Ecole polytech., Prague). Collection Czech. Chem. Commun. 15, 117-19 (1950) (in French).  
The test is based on the reduction of  $MnO_4^-$  by the substances named.  $MnO_4^-$  paper (I) is prepd. by moistening filter paper with 0.002 N  $KMnO_4$  for 30 min. and drying. Pieces 1 X 1 cm. are used. A 0.5% soln. (II) of benzidine in 10% HOAc is used to form a visible color. Place a micro drop (0.003 ml.) of test soln. on I and dip in II for a few sec. White spots on the blue paper show the presence of  $SO_4^{2-}$ ,  $SO_3^{2-}$ ,  $S^{2-}$ , or  $H_2S$ . HCHO causes green spots.  $Cl^-$ ,  $Br^-$ ,  $NO_3^-$ ,  $SO_4^{2-}$ ,  $F^-$ ,  $SiO_4^{4-}$ ,  $MoO_4^{2-}$ , and  $WO_4^{2-}$  cause a faint purple spot. The dilns. and sensitivities are  $H_2S$ : 1:3 X 10<sup>4</sup>, 0.1 [B] 0.003;  $SO_4^{2-}$ : 1:8 X 10<sup>4</sup>, 0.6 [B] 0.003;  $SO_3^{2-}$ : 1:1.5 X 10<sup>4</sup>, 0.2 [B] 0.003;  $S^{2-}$ : 1:3 X 10<sup>4</sup>, 0.1 [B] 0.003; and HCHO: 1:750, 4[B] 0.003.

K. G. Stote

CA

10

Metallic salts of salicylaldehyde thiosemicarbazone.  
V. Horváka and Z. Holzbecher. *Cahiers Carboch. Chem. Commun.*, 13, 267-74 (1950) (in French). *o*-HO-C<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>NNHCSNH<sub>2</sub> (I), because of its tautomeric possibilities, behaves as a mono- or diacid in the formation of cryst. metallic salts. I is prep'd. by mixing alc. *o*-HOCH<sub>2</sub>CHO with aq. H<sub>2</sub>NNHCSNH<sub>2</sub>. HCl in equimol. quantities, chilling, and recrystg. the pptd. I from hot EtOH

The metallic salts are prep'd. with a 0.6% alc. soln. of I and upon the addn. of NaOAc characteristic colors are formed. The following salts are prep'd.: C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>SNI-NH<sub>3</sub>, red-brown; C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>SPb, yellow; (C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>S)<sub>2</sub>Cu, yellow; C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>STl, yellow; C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>SCo<sub>2</sub>C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>SNH<sub>3</sub>, or C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>SCo<sub>2</sub>NH<sub>3</sub>; C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>S black; C<sub>6</sub>H<sub>5</sub>ON<sub>2</sub>SCu, brown-black. The structural formulas of the above salts are discussed.  
Bernard Klein

C.A.

Gravimetric determination of cadmium by using the thiosemicarbazone of salicylaldehyde. V. Hovorka and Z. Holzinger. Collection Czechoslov. Chem. Commun. 15, 275-80 (1950) (in French). Cd in solns. of  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  ions as the salt of thiosemicarbazone of salicylaldehyde  $\text{Cd}(\text{C}_6\text{H}_4\text{ONCS})_2$  (I) can be readily detd. within the limits of exptl. error. An excess of NaOAc does not affect the results of the detn. but the presence of  $\text{Cl}^-$ ,  $\text{F}^-$ , tartrate, and citrate prevent quant. pptn. Alkali earth metals are entrained in the ppt. A soln. contg. 0.004 to 0.14 g. Cd in 25 to 75 cc. is mixed with 45 to 50 cc. of 0.5% aq. soln. of 1.03 g. Cd. Ten % NaOAc is then added with stirring, and the soln. is heated to boiling until faint turbidity. The addn. is halted and the turbidity is cleared by heating until a yellow crystn. ppt. forms. The remainder of the NaOAc is added. Usually 20-30 cc. of NaOAc is required (0.03 g. Cd). The mixt. is warmed 0.6 hr. and chilled. The ppt. is collected on a crucible, washed with 80 to 150 cc. icewater and 5 to 10 cc. EtOH, dried at 110°, and weighed.  
Bernard Klein

C.A.

**Microchemical confirmation of manganese with biacetyl-oxime-thiosemicarbazone.** V. Hovorka and Z. Holzheigl. *Collection Czechoslov. Chem. Commun.*, 15, 287-291 (1950) (in French). - Microchem. confirmation of Mn<sup>++</sup> in solns. can be obtained either on a spot plate or paper impregnated with an alc. soln. of biacetyl oxime thiosemicarbazone. After addn. of NH<sub>3</sub> the spot plate or test paper shows either a red-violet ring or spot. Common ions do not interfere. Mn<sup>++</sup> can be confirmed in the presence of Fe<sup>++</sup> by means of tartate, or in the presence of Ni by the use of KCN. The reagent is prep'd. by heating to 80°/10 g. AcC(=NOH)Me, 17.2 g. AcONa in 500 ml. H<sub>2</sub>O acidified with 0.6 ml. AcOH, and adding 500 ml. of a warm aq. soln. contg. 0.8 g. H<sub>2</sub>NNHC(SNH)<sub>2</sub>. A 90% yield of a pale yellow solid is soon deposited and after heating for an addnl. 2 hrs. is chilled for several days and recrystd. from EtOH, in. 200° (decomp.). Alc. solns. are colorless and unstable, turning brown after several days.      Bernard Klein

C.R.  
1951

Organic Chemistry  
10

**Metallic salts of diacetyl oxime thiosemicarbazone.** V  
Hovorka and Z. Holubecler (Ecole polytechnique, Prague)  
*Collection Compt. Rend. Commun.*, 15, 437 (1930) (in French);  
cf. C.I. 44, 2075; 45, 973. It is concluded that there are 2 tautomeric forms of diacetyl oxime thiosemicarbazone:  $\text{HON}(\text{CMe}_2\text{CMe}_2)\text{NNHCNSN}$ ; (I) and  
 $\text{HON}(\text{CMe}_2\text{CMe}_2\text{NNHC})\text{NHSH}$  (II). I is a mono base,  
II a dibasic acid. Derivs. of I: *Cr salt*,  $(\text{C}_6\text{H}_5\text{ON})_2\text{Cr}_2\text{O}_7$ , brown;  
*Cd salt*,  $(\text{C}_6\text{H}_5\text{ON})_2\text{CdH}_2\text{O}$ , yellow.  
Derivs. of II: *Ni salt*,  $\text{C}_6\text{H}_5\text{ON}(\text{SNCu})_2$ , brown-red; *Co salt*,  
 $\text{C}_6\text{H}_5\text{ON}(\text{SCu})_2$ . The *fern salt*,  $\text{C}_6\text{H}_5\text{ON}(\text{FeCl})_2\text{ON}$   
seems to contain one group each of I and II. It occurs in a  
brownish-yellow and in a black form, which give the same  
Debye diagram.  
Alfred Hoffmann

*Brit abs*

Radiant burner. N. Holzbecker (Patent, 1960, 30, 196-191).—A considerable part of the radiance from Bunsen burner-burners is directed away from the axis of the reflector. A porous diaphragm radiating according to Lambert's law may be adapted to a Bunsen-type burner, and is probably the best design for a radiant burner.  
J.R. Tuncer.

CA

17

*3-Semicarbazones and 3-(3-thiosemicarbazones) of isatin and its derivatives.* Z. Holzbecher (Tech. Univ., Prague). *Chem. Listy* 44, 126-7 (1950).—To 20 g. isatin suspended in 1 l. boiling water was added 17 g.  $H_2NNHCONH_2 \cdot HCl$  (I) in  $H_2O$ , the mixt. boiled a few min., and the ppt. filtered by suction and washed with water to yield 27 g. (97%) isatin *3-semicarbazone* (II), yellow needles, decomp.  $260^\circ$ , solv. in  $MeOH$  1% at the b.p., 0.5% at  $20^\circ$ , less sol. in  $EtOH$  and  $Me_2CO$ . The *Na salt* (III) of II was prep'd. by adding an equiv. amt. of  $NaOH$  to II in aq.  $EtOH$ ; its solv. is 1 g. in 100 ml. 25%  $EtOH$ . *1-Methylisatin 3-semicarbazone* was prep'd. analogously from 10 g. 1-methylisatin and 7.5 g. I. Two forms of crystals were isolated; after drying at  $110^\circ$ , both forms decom.  $230^\circ$  (from water). The solv. in  $H_2O$ ,  $MeOH$ ,  $EtOH$ , and  $Me_2CO$  is less than 0.1%. The solv. of the *Na salt* in 50%  $EtOH$  is 1% at  $20^\circ$ . To 15 g. 1-benzylisatin in 300 ml. hot  $EtOH$  was added 8 g. I in 50 ml. water, and the ppt. filtered by suction, washed with hot water and  $EtOH$ , and dried at  $110^\circ$  to yield 17.5 g. (plus 0.6 g. from the mother liquor) (97%) *1-benzylisatin 3-semicarbazone*, m.  $214^\circ$  (decompn. from  $EtOH$ ), solv. in boiling  $AcOH$  and in 0.2%  $NaOH$  in 4:6  $EtOH \cdot H_2O$  1%. In  $MeOH$ ,  $EtOH$ ,  $AmOH$ ,  $Me_2CO$ , and  $C_4H_9N$  less than 0.5%, insol. in  $C_6H_6$ ,  $CS_2$ , and  $CHCl_3$ . *Isatin 3-(3-thiosemicarbazone)*, m.  $255^\circ$  (decompn. from  $MeOH$ ), was prep'd. from 15 g. isatin in 1 l. boiling water and 11 g.

$H_2NCSNH_2 \cdot H_2O$  (IV) in almost 100% yield. The solv. in  $EtOH$  is 0.5% in  $MeOH$  and in 0.2%  $NaOH$  in 1:1  $EtOH$ : $1\%$ , and in acetone more than 2% at room temp. IV (4.5 g.) in  $H_2O$  was added to 5 g. 1-methylisatin in 300 ml. boiling water and boiled 1 hr.; the addn. of 0.6 ml. dil.  $HCl$  accelerated the reaction. The ppt., washed with water and dried at  $110^\circ$ , yielded 7.1 g. (98%) *1-methylisatin 3-thiosemicarbazone*, m.  $210^\circ$  (decompn. from  $MeOH$ ). Recryst. IV was necessary. The solv. is 1% in boiling, 0.5% in cold  $Me_2CO$ , less than 0.5% in  $MeOH$ ,  $EtOH$ , and  $Et_2O$ ;  $AcOH$ , and 1% in 0.3%  $NaOH$  in  $H_2O$ : $EtOH$  (3:7). IV (5 g.) in 30 ml. hot water and 0.6 ml.  $HCl$  (1:1) added to 12 g. 1-benzylisatin in 250 ml. boiling  $EtOH$ , and the mixt. boiled a few min., yielded 18.2 g. (97%) *1-benzylisatin 3-thiosemicarbazone*, m.  $240^\circ$  (decompn.), solv. in boiling  $C_4H_9N$  1%, in  $MeOH$ ,  $EtOH$ ,  $AmOH$ ,  $Me_2CO$ , or  $AcOH$  less than 0.5%, in 0.2%  $NaOH$  in  $H_2O$ : $Me_2CO$  (1:3) 1%.

M. Hudlický

(A) HOLZBECHER, Z.

*Holzbecher's Remarks*

Reactions of phenylboric acid and its nitro derivatives with metal salts. Záviš Holzbecher (Tech. Univ., Prague, Czech. A. Chem. Listy 46: 17-19 (1942); Phil. Mag. (1)

ppm. CuO) from Cu(OAc)<sub>2</sub> soln. Cu(OAc)<sub>2</sub> and *c*-NO<sub>2</sub>-C<sub>6</sub>H<sub>4</sub>BO<sub>3</sub>, or *m*-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>BO<sub>3</sub>, give CuO and 2,3-dinitrodiphenyl (60%) or 3,3'-dinitrodiphenyl (40%), resp. Hg(NO<sub>3</sub>)<sub>2</sub> and Li in 10% NaOAc ppm. 100% Pb<sub>2</sub>Hg, m. 123°. I opt. Phil. Mag. 20: 67, in 100% yield from a soln. containing Hg(NO<sub>3</sub>)<sub>2</sub>, KI, NH<sub>4</sub>NO<sub>3</sub>, and NH<sub>3</sub>. M. Hudlický

CA HOLZBECHER, Z.

Gravimetric estimation of mercury with phenylboric acid.  
Zavis Holzbecher (Tech. Univ., Prague, Czech.). Chem.  
Listy 46, 593 (1952).—Phenylboric acid (I) ppts. Hg as  
Ph<sub>3</sub>Hg. The analysis is carried out in a soln. buffered with  
NaOAc soln. in the presence of NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, tartate or  
citrate, or in an ammoniacal soln. if Cl<sup>-</sup>, Br<sup>-</sup>, or CNS<sup>-</sup> is  
present. Procedure: To 10-75 ml. of soln. contg. 0.005  
0.1 g. Hg add 5-25 ml. 1% soln. of I and 5-50 ml. of 10%  
NaOAc. Filter, wash with cold, satd. soln. of I into a pot.  
Filter, wash with cold, satd. soln. of I into a pot.  
Finally wash with two 5-ml.  
portions of H<sub>2</sub>O. Dry at 70° and weigh in a vacuum.  
In the presence of halide anions, use concd. NH<sub>4</sub>OH instead of NaOAc.  
M. Hudlický

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Metal salts of the thiosemicarbazones derived from 2,6-diformyl and 2,6-ethyl ester and from 2-acetyl-6,6-diformyl-2,6-diformyl-2,6-dihydro-4H-1,3-thiodiazin-4-one (I) were obtained by reaction of the corresponding thione with metal chlorides in benzene. The product of the reaction of I with zinc chloride is a yellow solid, m.p. 160°C. It is soluble in benzene and in  $\text{CH}_2\text{Cl}_2$ , but insoluble in  $\text{CHCl}_3$ . The infrared spectrum of II shows absorption bands at 3350, 1650, 1550, 1450, 1350, 1250, 1150, 1050, 950, 850, 750, and 650 cm<sup>-1</sup>. The ultraviolet spectrum of II shows absorption bands at 3350, 1650, 1550, 1450, 1350, 1250, 1150, 1050, 950, 850, 750, and 650 cm<sup>-1</sup>.

The structure of II was determined by comparison with the structure of IV.



HOLZBECHER, Z.

"New Fluorescent Reactions of Aluminum" p. 680, (CHEMICKÉ LISTY, Vol. 47, no. 5, May 1953, Praha, Czechoslovakia).

SO: Monthly List of East European Accessions, LC, Vol. 2, No. 11, Nov. 1953, Uncl.

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HOLZBECHER, Z.

Metal salts of the thiosemicarbazones derived from benzoylformic acid and its ethyl ester, and from acetophenone [in Russian with summary in English]. Sbor.Chekh.khim.rab. 19 no.1:69-76 F '54. (MLRA 7:6)

1. First Department of Analytical Chemistry, Technical University, Prague. (Semicarbazones) (Glyoxylic acid) (Acetophenone)

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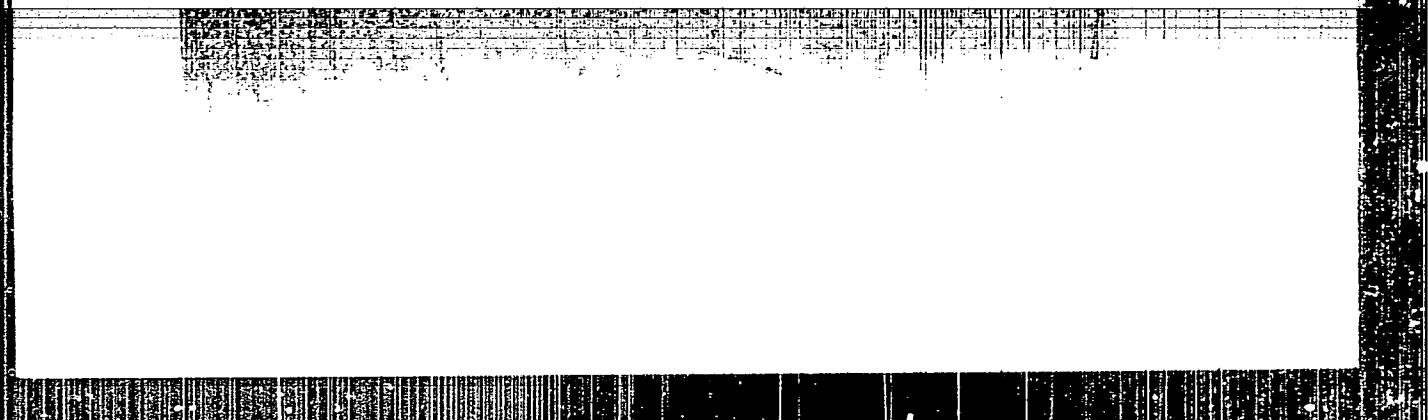
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Holzbecher, Z

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G. Z. J. (1)

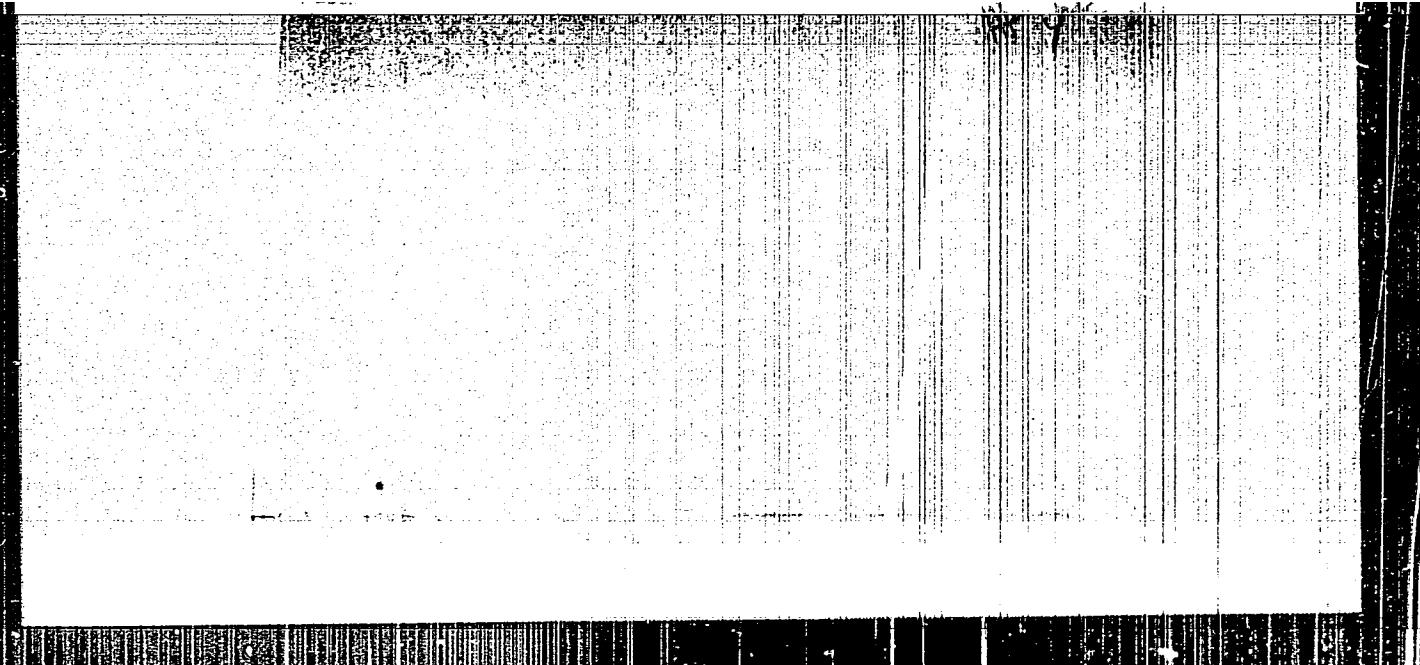
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CIA-RDP86-00513R000618120003-0

HOLZBECKER L

Fluorescence Detection and determination  
of nucleic acids with 2,6-dihydroxyphenylimidazoles

APPROVED FOR RELEASE: 09/21/2001

CIA-RDP86-00513R000618120003-0"

HOLZBECHER, Z.

CZECH

Now detection and determination of zinc by means of fluorescence. Z. Holzbecher (Wiss. Beitr. Chem., 19, 884-92 (1955)). Benzylalcohol,  $\beta$ -naphthol,  $\alpha$ -naphthol,  $\beta$ -naphthoquinone (I), and methylbenzene-carboxylic acid esters are suitable reagents for detection and determination of  $Zn^{+2}$  by hydride are suitable reagents for detection and determination of  $Zn^{+2}$  by means of fluorescence in ultraviolet light. The detection is possible in the range of  $\beta$ -naphthol-Zn in 0.1 ml. soln., even in the presence of other elements. Li, Na, K, NH<sub>4</sub>, Sr, and Ba do not interfere up to doses, 500  $\mu$  per 50 ml. Al and Cu must be screened by adding a 3% soln. of NaOH. To 10 ml. Zn in  $Zn(OAc)_2$ , add to a 5-ml. sample contg. up to 1.3 mg. Zn, 10 ml. 0.1M acetate buffer, 10 ml. EtOH, 0.5 ml. 0.01M I, 0.01M EtOH, dil. to 50 ml., and measure the fluorescence.

CZECH

Fluorescence spectra of salicylaldehyde condensation products and their salts. III. Acetyl hydrazone, semicarbazone, and thiosemicarbazone of salicylaldehyde.

Zdenek Hrdlicka (Vysoká škola chem.-technol., Praha).

Czechoslovakia 49, 1102-6(1958); cf. C.A. 52, 12971c.

Fluorescence and absorption spectra of acetyl hydrazone (I), semicarbazone (III), and thiosemicarbazone (III) of

salicylaldehyde are given in the range of 330-630 m $\mu$  for the reagents and their Zn and Al salts. From the effect of pH on the intensity of fluorescence the following dissociation constants were detd.: I  $5 \times 10^{-4}$ , II  $3 \times 10^{-4}$ , III  $4 \times 10^{-4}$ . The compn. of the salts of I-III in aq.-alc. solns., as detd. by the method of continuous variations, corresponds to the ratio 1 Zn : 1 mole of the reagent I-III. The ratio of Al to Zn found to be 1:2. The inner-complex formulas for the

I-III-Zn salt. M. Hrdlicka

~~HOLZBECHER, Zavis~~

CZECHOSLOVAKIA / Physical Chemistry. Molecule. Chemical Bond.

B-4

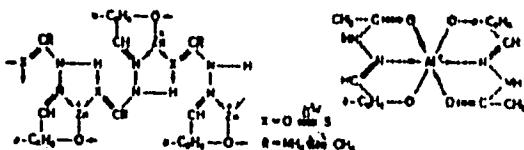
Abs Jour : Ref Zhur - Khimiya, No 8, 1957, 25771

Author : Zavis Holzbecher

Title : Fluorescence Spectra of Condensation Products of Salicylaldehyde and Its Salts. III. Acetylhydrazone, Semicarbazone and Thiosemicarbazone of Salicylaldehyde.

Orig Pub : Chem. listy, 1955, 49, No 8, 1162-1166; Sb. chekosl. khim. rabot, 1955, 20, No 6, 1297-1301

Abstract : The fluorescence spectra (F) and the absorption spectra of aqueous-alcohol (70%) solutions of acetylhydrazone (I), semicarbazone (II) and thiosemicarbazone (III) of salicylaldehyde and their Zn and Al salts were studied in the range from 330 to 560 m $\mu$ . The dependence between the intensity of F



Card 1/2

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CZECHOSLOVAKIA / Physical Chemistry, Molecule, Chemical Bond;

B-4

Abs Jour : Ref Zhur - Khimiya, No 8, 1957, 26771

Abstract : of the solutions in the visible spectrum part and the absorption intensity in the range from 350 to 370 mu was revealed. The maximum F is characteristic of alkaline solutions of I, II and III, and the minimum F is characteristic of acid solutions. The composition of Zn salts of I, II and III (1 : 1) and of Al salt of I (1 Al atom : 2 I molecules) was established by measuring the F spectra. The composition of Al salts of II and III is assumed to be the same. Intracomplex chain polymeric cations are characteristic of Zn salts. The acidity constants of I, II and III were computed from the dependence of F on pH by the method described earlier (RZhKhim, 1956, 13388). I ( $5 \pm 1$ ). $10^{-9}$ , II ( $3 \pm 1$ ). $10^{-9}$ ) mu( $4 \pm 1$ ). $10^{-7}$ .

For part II, see RZhKhim, 1956, 46031.

Card : 2/2

- 22 -

HOLZBECHER, Z.

"New fluorescent indicators. I. Neutralization titration. II. Volumetric determination of aluminum and zinc."

p. 425 (Chemicke Listy, Vol. 52, no. 3, 1958, Praha, Czechoslovakia)

Monthly Index of East European Accessions (EEAI) LC, Vol. 7, no. 9,  
September 1958

Country : Czechoslovakia  
 Category : Analytical Chemistry - General

E-1

Abs. Jour. : Ref Zhar-Khimiya, No 6, 1959

19048

Author : Holzbecher, ZAVIS

Institut. : Title : New Fluorescent Indicators. I. Titrations by the Neutralization Method.

Orig. Pub. : Chem. listy, 1958, 52, No 3, 425-429

Abstract : The authors have studied the products of condensation of salicylaldehydes (I): o-hydroxyphenyl-benzothiazole (II), o-hydroxyphenyl-benzoxazole (III), o-hydroxyphenyl-benzimidazole (IV), semicarbazone of I (V), acetyl-hydrazone of I (VI), thiosemicarbazone of I (VII), oxime of I (VIII), salicylidene-o-aminophenol (IX), salicylidene-m-aminophenol (X), salicylidene-p-aminophenol (XI), salicylidene-o-anisidine (XII), salicylidene-aniline (XIII) and salicylidene-semioxamzone (XIV) -- and have found that compounds which are suitable as fluorescent indicators, in titrations by the neutralization method in ultraviolet light, are only II, III, IV, V, VI and VII, which, in an alkaline medium, exhibit a

Card: 1/6

Country : Czechoslovakia  
 Category : Analytical Chemistry - General

E-1

Abs. Jour. : Ref Zhar-Khimiya, No 6, 1959

19048

Author :  
 Institut. :  
 Title :

Orig. Pub. :

Abstract : relatively strong blue- or blue-violet fluorescence. Intensity and color of fluorescence (F) of  $10^{-5}$  M solutions of all the derivatives of I under study were determined in citrate-, phosphate-, and borate-buffer solutions depending upon pH; changes in fluorescence and dissociation constants of these derivatives were determined from photometric or titrimetric data. II, III, IV, V, VI, and VII, were utilized as indicators in titrations of bases with strong and weak acids with strong bases; an alkaline medium the F of indicators is decreased on

Card: 2/6

E-3

Country : Czechoslovakia  
Category : Analytical Chemistry - General

E-1

Abs. Jour. : Ref Zhur.-Khimiya, No 6, 1959

19048

Author :  
Institut. :  
Title :

Orig Pub. :

Abstract : irradiation with ultraviolet light. Intensity of F of the indicators used was determined by means of a Duboscq colorimeter and a PRK-4 mercury-quartz lamp (220 v) as the source of ultraviolet radiation (with a VG-4 black light-filter, maximum transmission at 3660 Å). For individual indicators are listed the following: acid dissociation constants  $pK_x$ , as determined photometrically and by calculation; dissociation constants of bases  $pK_z$ ; color of F in alkaline medium; maximum F-intensity of a  $10^{-5}$  M solution of the reagent, expressed in % on the basis of F-intensity of  $10^{-5}$  M solution of quinine sulfate in 0.1 N  $H_2SO_4$  (blue light-filter, maximum transmission at 4450 Å), which is taken

Card: 3/6

Country : Czechoslovakia  
Category : Analytical Chemistry - General

E-1

Abs. Jour. : Ref Zhur-Khimiya, No 6, 1959

15048

Author :  
Institut. :  
Title :

Orig. Pub. :

Abstract : as equal to 100% (in the case of II, III, IV, V, VI, VIII), or on the basis of F-intensity of  $10^{-5}$  M solution of Na-salt of fluorescein in 0.1 N NaOH (light-filter 5250 Å) taken as equal to 50% (in the case of VII, IX, X, XI, XII, XIII, XIV). These data are as follows: II 9.3, 10.10, >11, blue-green, 500; III 9.3, 11.05, >11, blue-violet, 170; IV not determined, 9.90, 9.05, blue-violet, 150; V 8.5, 9.25, >11, light-azure, 50; VI 8.3, 9.60, >11, greenish-blue, 100; VII 8.4, 9.20, >11, blue-green, 30; VIII 9.7, 9.65, >11, blue-green, 5; IX 7.8, 8.63, 9.45, yellow-green, 2.5; X 7.9, 8.75, 9.85, green, 2.0; XI 8.3, 8.85, 8.75, green, 2.5;

Card: 4/6

E-4

Country : Czechoslovakia  
Category : Analytical Chemistry - General

E-1

Abs. Jour. : Ref Zhur-Khimiya, No 6, 1959

19048

Author :  
Institut. :  
Title :

Orig Pub. :

Abstract : XII 8.2, 8.60, 9.80, green, 2.5; XIII 7.9, 8.70,  
4.25, yellow-green, 2.0; XIV 7.2, 9.35, >11, greenish-blue,  
0.5. The lower values of photometrically determined  $pK_a$  are  
caused by partial decomposition of reagents in alkaline  
medium on ultraviolet irradiation, and quenching of F by  
some anions present in buffer solutions. On titrimetric  
determination of acids ( $HCl$ ,  $H_2SO_4$ ,  $HNO_3$ ,  $CH_3COOH$ ,  $H_3PO_4$ ) a  
simple arrangement was used for lateral illumination of the  
solution being titrated and observation of fluorescence  
from above. To 25 ml of the solution were added 1-2 drops of  
0.1% alcohol-solution of indicator (II, III, IV, V, VI, VII)  
after which titration was carried out with 0.1 N NaOH until  
Card: 5/6

Country : Czechoslovakia  
Category : Analytical Chemistry - General E-1  
Abs. Jour. : Ref Zhur-Khimiyu, No 6, 1959 19048

Author :  
Institut. :  
Title :

Orig. Pub. :

Abstract : a faint F developed; reaching of equivalence point was checked by comparison with control experiment solution. The results thus obtained are somewhat too high in comparison with those obtained using methyl orange or phenolphthalein, but are within the limits of permissible error ( $\pm 0.2\%$ ). For an accurate determination of  $\text{CH}_3\text{COOH}$  only II, III and IV can be used; for titration of 2nd  $\text{H}^+$  of  $\text{H}_3\text{PO}_4$  only III is suitable. -- J. Vanecek.

Card: 6/6

E-5

Country	: Czechoslovakia	E-2
Category	: Analytical Chemistry - Analysis of Inorganic Substances	
Abs. Jour.	: Ref Zhur-Khimya, No 6, 1959	19079
Author	: Holzbecher, Z.	
Institut.	:	
Title	: New Fluorescent Indicators. II. Titrimetric Determination of Aluminum and Zinc.	
Orig Pub.	: Chem. listy, 1958, 52, No 3, 430-438	
Abstract	Formation of the intensively fluorescent, in ultraviolet light, Al-salt of salicyliene-o-aminophenol, (I) (green-yellowish fluorescence) and Zn-salt of acetylhydrazone of salicylaldehyde (II) (blue fluorescence), is utilized in the development of a new method of titrimetric determination of Al and Zn, singly and in admixtures with other numerous elements. Of the investigated titration agents best suited were found to be NaF for Al and Complexon III for Zn. In determining Al in $KAl(SO_4)_2$ , there are added to 10 ml of solution to be analyzed (20-40 mg Al), at pH 4-5, 10 ml of acetate buffer solution of pH about 5.1 (1 volume	
Card:	1/6	

Country :	Czechoslovakia	E-2
Category:	Analytical Chemistry - Analysis of Inorganic Substances	
Abs. Jour.:	Ref Zhur-Khimiya, No 6, 1959	19079

Author :  
Institut. :  
Title :

Orig. Pub. :

Abstract : of 1 M  $\text{CH}_3\text{COOH}$  + 2 volumes of 1 M  $\text{CH}_3\text{COONa}$ ) and 2 ml 0.05% alcohol solution of indicator I (solutions of I are prepared anew every week) and titration is carried out with 0.6 M solution of NaF under ultraviolet light (communication 1, RZhKhim, 1959, 19048). For each sample at least two determinations are made and the 1st titrated solution is utilized as control in the second titration in which almost the entire required amount of NaF solution is added rapidly and at once, after which titration is completed under ultraviolet light. 1 ml of 0.6 M solution of NaF corresponds to  $2.715 \pm 0.008$  mg Al (average of 5 determinations). To secure

Card: 2/6

E-11

Country	: Czechoslovakia	E-2
Category	: Analytical Chemistry - Analysis of Inorganic Substances	
Abs. Jour.	: Ref Zhur-Khimiya, No 6, 1959	19079

Author :  
Institut. :  
Title :

Orig Pub. :

Abstract : correct results it is necessary to follow exactly the described procedure as concerns conditions of titration, volumes and concentrations of the solutions. 20-40 mg Al in 10 ml solution can be determined with a maximum error of  $\pm 1\%$ ; with a concentration of about 26 mg Al in 10 ml the relative error of determination is  $\pm 0.3\%$ . Determination of Al is not interfered with by the presence of 100-350 mg Na<sup>+</sup>, K<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, Li<sup>+</sup>, Tl<sup>+</sup>, Ag<sup>+</sup>, Cd<sup>2+</sup>, Ni<sup>2+</sup>, Co<sup>2+</sup>, Zn<sup>2+</sup> and Mn<sup>2+</sup> and also of 10-35 mg Pb<sup>(2+)</sup>, Hg<sup>(2+)</sup>, As<sup>(3+)</sup>, P<sup>(5+)</sup>, As<sup>(5+)</sup>, U<sup>(6+)</sup> and Cr<sup>(3+)</sup>. Lower results are obtained (due to quenching of fluorescence) in the presence of even

Card: 3/6

Country : Czechoslovakia  
Category : Analytical Chemistry - Analysis of  
Inorganic Substances  
Abs. Jour. : Ref Zhur-Khimiya, No 6, 1959

E-2

19079

Author :  
Institut. :  
Title :

Orig. Pub. :

Abstract : small amounts of molybdates, tartrates,  $\text{Fe}^{3+}$ , and of large amounts of  $\text{Cr}^{3+}$  and  $\text{U}^{(6+)}$ . Titration of Al can not be carried out in the presence of  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Sr}^{2+}$ ,  $\text{Ba}^{2+}$ ,  $\text{Ti}^{4+}$ ,  $\text{Ce}^{3+}$ ,  $\text{Th}^{4+}$ ,  $\text{Zr}^{4+}$ ,  $\text{La}^{3+}$ ,  $\text{Be}^{2+}$ , Sb and Sn. By addition of 20% solution of  $\text{Na}_2\text{S}_2\text{O}_3$ , immediately before titration, it is possible to eliminate the interfering effect of small amounts of Bi, Cu and Pb. If a small amount of  $\text{Fe}^{3+}$  is present its interfering effect can be eliminated by reduction with thiosulfate in the presence of  $\text{Cu}^{2+}$ -salt which speeds up the reduction. On determining Zn in  $\text{ZnSO}_4$ , 10-20 ml of the almost neutral solution (6-170 mg Zn) are mixed with 10 ml

Card: 4/6

E-12

E-2

Country : Czechoslovakia  
Category : Analytical Chemistry - Analysis of  
            Inorganic Substances  
Abs. Jour. : Ref Zhur-Khimiya, No 6, 1959

19079

Author :  
Institut. :  
Title :

Orig Pub. :

Abstract : acetate buffer solution and 2 ml 0.1% alcohol solution of indicator II (the solution of II is stable for one year), and titrated with 0.1 M solution of Complexon III to a quenching of the fluorescence in ultraviolet light; a control experiment is run concurrently. Relative error of determination of 30-160 mg Zn in 5-25 ml varies from - 0.16 to + 0.45%. Determination of Zn is not interfered with by the presence of 50-400 mg Na<sup>+</sup>, K<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, Li<sup>+</sup>, Tl<sup>+</sup>, Ag<sup>+</sup>, Sr<sup>2+</sup>, Ba<sup>2+</sup>, and also of thiosulfates, fluorides, tartrates, small amounts of arsenites, arsenates, and phosphates. The presence of large amounts of phosphates, arsenates, and of Sb and Sn in the form of tartrate complexes, causes inaccurate results.

Card: 5/6

E-15

HOLZBECHER, Z.

COUNTRY : Czechoslovakia E-2  
CATEGORY : Analytical Chemistry. Analysis of Inorganic  
Substances.

ABS. JOUR. : AZKhim., No. 19, 1959, No. 67724

AUTHOR : Holzbecher, Z.

JOUR. :

TITLE : New Fluorescent Indicators. III. Titrimetric  
Determination of Aluminum and Zinc in Alloys

ORIG. PUB. : Chem. listy, 1958, 52, No 9, 1822-1825

ABSTRACT : Description of a rapid titrimetric method for  
determining Al and Zn in Cu- and Mg-alloys, and also Al in  
Zn-alloys. The sample being analyzed is dissolved in dilute  
HNO<sub>3</sub> (in the case of Cu- and Mg-alloys) or in dilute HCl  
(in the case of Zn-alloys), the interfering components are  
removed and titration is carried out in ultraviolet light:  
Al<sup>3+</sup> with 0.6 M solution of NaF to the fluorescent indicator  
methylisobutyl-o-aminophenol, and Zn<sup>2+</sup> - with 0.1 M solution  
of C<sub>12</sub>H<sub>22</sub>O<sub>11</sub> III, to the acetyl hydrogen of malonilic  
acid indicator, until the yellow-green and blue fluorescence,  
respectively, disappears. The titration conditions  
must be buffered with an acetate buffer solution (pH 5.5).

SABD: 1/4

COUNTRY	:	Czechoslovakia	E-2
CATEGORY	:		
ABSTRACT JOUR.	:	RZKhim., No. 19, 1959, No. 7724	
AUTHOR	:		
INST.	:		
TITLE	:		
ORIG. PUB.	:		
ABSTRACT	:	<p>Titration of Al is interfered with by Sn, Cu, Fe, and by large amounts of Pb and Zn, while the titration of Zn, in addition of the above stated elements, is also interfered with by Al, and by large amounts of Ni and Mn. Separating Cu-alloys (brass, bronze) Al is removed in <math>\text{AlCl}_3</math> on dissolution of the sample in <math>\text{HNO}_3</math>, and the remaining Sn is removed electrolytically; the remaining Cu, together with a small amount of Fe (or Pb) is masked by means of <math>\text{Na}_2\text{B}_4\text{O}_7</math>. Before the titration of Zn, Al is masked with an excess of NaF. In analyzing samples containing large amounts of Mn and Ni, Zn is separated beforehand as ZnS. In analyses of Ag-alloys the solution of CARE: 2/4</p>	

COUNTRY : Czechoslovakia  
CATEGORY :  
AES. JOUR. : RZKhim., No. 19, 1959, No. 67724

B-2

REF ID: A  
TITLE :

ORIG. PUB. :

ABSTRACT : If the sample is evaporated with  $HgCl_2$  and the residue is filtered off; Cu is separated as a precipitate of hexamethylbenzidine, and the Al (in excess amounts of the solution) -- as basic; the separated aluminum precipitates are dissolved in HCl and Al, or Zn, are reprecipitated as described above. Cu determination of Al in samples which do not contain large amounts of Ni, the primary separation of Al is not indispensable and it is usually sufficient to cover Cu and Fe (or Fe) with hexamethylbenzidine. The sample is rejected depending upon content of Cu and methods of its determination.

CARD: 3/4

HOLZBECHER, Z.

"Ionic equilibria in analytical chemistry" by H. Freiser, Q.  
Fernando. Reviewed by Z. Holzbecher. Chem listy 58 no.6:995-  
996 Ag '64

HOLZBECHER, Z.

Fluorescence of metallic salts of condensation products of  
salicylaldehyde. Coll Cs chem 25 no.12:3915-3919 '59.  
(EEAI 9:6)

1. Institut fur analytische Chemie, Technische Hochschule fur  
Chemie, Prag.  
(Fluorescence) (Salts) (Salicylaldehyde)

HOLZBECHER, Z.

Fluorescence of metal chelates of resorcyaldehyde and its derivatives.  
Coll Cz Chem 25 no.4:977-982 Ap '60. (ESAI 9:12)

1. Institut fur analytische Chemie, Technische Hochschule fur Chemie,  
Prag. (Fluorescence) (Chelatometry) (Resorcyaldehyde)

HOLZBECHER, Z.

Test of fluorescence of scandium, gallium and zirconium on paper-chromatograms by means of formylhydrazone of resorcyldialdehydes.  
Coll Cz Chem 26 no.4:1204-1206 Ap '61.

1. Institut fur analytische Chemie, Technische Hochschule fur Chemie,  
Frag.

(Scandium) (Gallium) (Zirconium)

S/081/62/000/001/014/067  
B156/B101

AUTHOR:

Holzbecher, Z.

TITLE:

Scandium, gallium and zirconium detected by fluorescence

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 1, 1962, 138, abstract  
1D35 (Acta chim. Acad. scient. hung., v. 27, 1961, nos. 1-4,  
413-416)

TEXT: Sc, Ga and Zr have been detected by means of the fluorescence in UV light exhibited by the complexes which they form with the products of condensation (CP) between salicylic or resorcylic aldehydes and aromatic amines or hydrazine derivatives. The relationship of intensity of fluorescence to the ratio between the cation charge and radius, and to the pH of the solution, has been studied for a series of complexes with several CP. Chromatographic recordings have been made of a mixture of Sc, Ga and Zr on Whatman paper no. 1 by means of a system of solvents ( $n\text{-C}_4\text{H}_9\text{OH}$  -  $\text{CH}_3\text{COOC}_2\text{H}_5$  - conc. HCl); the chromatograms are dried and treated with a

Card 1/2

Scandium, gallium and zirconium ...

S/081/62/000/001/014/067  
B156/B101

0.01% ethanol solution of formylhydrazone of resorcylic aldehyde, and zones examined in UV-light ( $R_f$  Sc 0.10-0.19, Ga 0.70-0.97, Zr 0.0-0.1). If the chromatogram is treated with  $NH_3$ , zones of Al, Zn and Be are also revealed in UV-light. The method enables up to 0.04 γ Sc, 0.06 γ Ga and 0.5 γ Zr to be determined in the presence of almost all other cations. [Abstracter's note: Complete translation.]

Card 2/2

HOLZBECHER, Z.; PULKRAB, P.

Fluorometric determination of aluminum by means of formyl  
hydrazone of salicylaldehyde. Coll Cs Chem 27 no.5:1142-1149  
My '62.

1. Institut fur analytische Chemie, Technische Hochschule fur  
Chemie, Prag.